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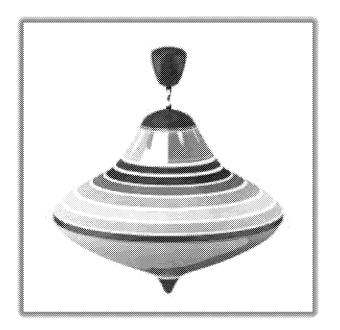
# **TestAmerica**

THE LEADER IN ENVIRONMENTAL TESTING

# Closing the PFAS Mass Balance: The Total Oxidizable Precursor (TOP) Assay

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### TOP Assay - Outline



#### Introduction to PFASs

What are PFASs?

Formation/Toxicity and Risk

Regulatory Review

### The TOP Assay

Background

What is the TOP assay

How does it work?

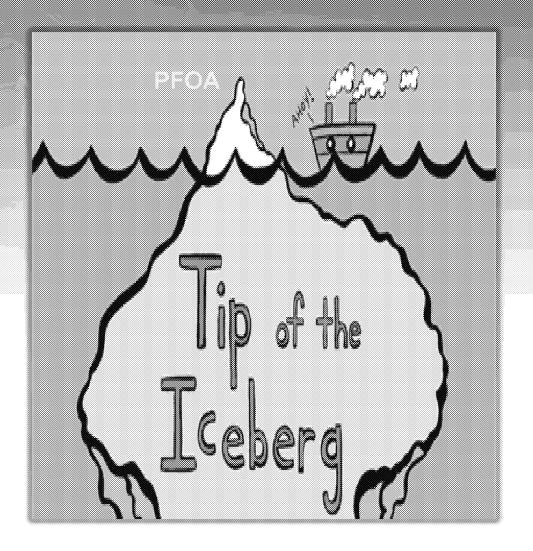
The chemical reaction

What do the results mean?

What are the limitations?

**Future Concerns** 

Capabilities and Questions?



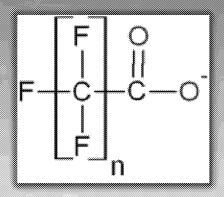
### Briefly - What are PFASs?

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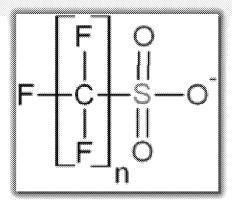
- Class of synthetic compounds containing carbon chains with fluorine attached to these chains.
- The C-F bond is the shortest and the strongest bond in nature.
- PFC Subset of PFAS completely fluorinated compounds. PFOS and PFOA are PFCs (no hydrogen atoms)
- PFAAs Perfluoroalkyl acids 2 classes PFCAs and PFSAs



PFAS do not degrade BUT they do biotransform



Perfluoroalkyl Carboxylate



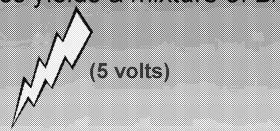
Perfluoroalkyl Sulfonate



### **PFAS Formation**

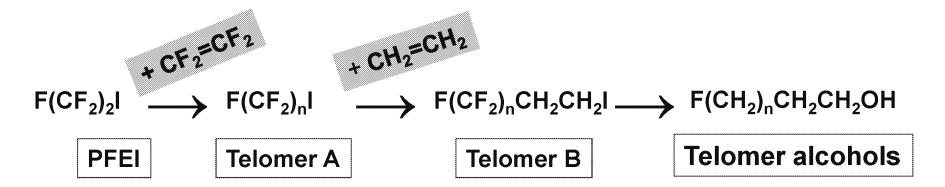


ECF Reaction: Process yields a mixture of B/L isomers



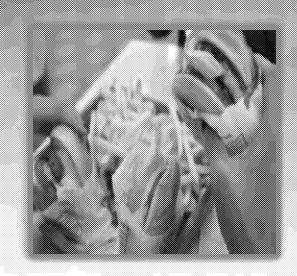
 $C_8H_{17}SO_2F + 34HF \rightarrow C_8F_{17}SO_2F + 17H_2 (POSF)$ 

➤ **Telomer Reaction:** Process yields 100% linear isomers (Synthesis of building blocks leading to fluorotelomer alcohols)



### Exposure, Toxicity and Risk







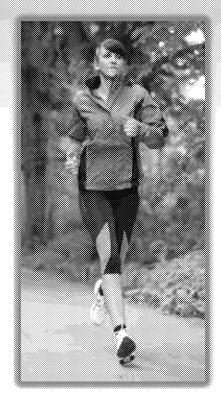
- Major source of non-occupational exposure to humans is from food and air (predominately fish consumption)
- Human and wildlife exposure can continue even though the chemicals are no longer in use, due to persistence.
- PFOS and PFOA have half-lives in humans ranging from 2 to 9 years, depending on the study.
- PFOA associated with liver, pancreatic, testicular, and mammary gland tumors in laboratory animals. PFOS causes liver and thyroid cancer in rats
- PFOA and PFOS are likely carcinogenic in humans. Pathways are being studied.

## PFAS - Regulatory Timeline



| When      | Who    | What Happened  |
|-----------|--------|--|
| 1980s     | EU     | Groundwater directive to prevent discharge of PFOS   |
| 2002      | US EPA | Initiated voluntary phase out of PFOS  |
| 2002      | ЗМ     | Discontinued making PFOS (7 other makers complied)   |
| 2006      | US EPA | Announced 2010 (95%)/15(100%) PFOA Stewardship Program   |
| 2008      | Canada | Regulated and prohibited PFOS imports to Canada  |
| 2009      | UN     | Stockholm Convention - adds PFOS to Annex B  |
| 2010      | US EPA | 2010 PFOA Stewardship program - must reduce PFOA use by 95%                                    |
| 2015      | US EPA | Must 100% eliminate the use of PFOA by December 31,2015.                                       |
| May 2016  | US EPA | PFOS and PFOA life time health limits reduced to 70 ppt each or the total if both are present. |
| Sept 2016 | NJ     | DWQI proposed PFOA drinking water MCL of 14 ppt  |



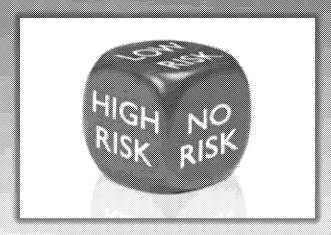


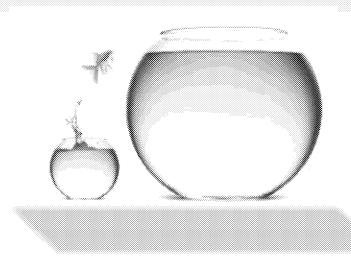
What started happening in 2016?

# What is Risk? Why Do We Care About PFOA?



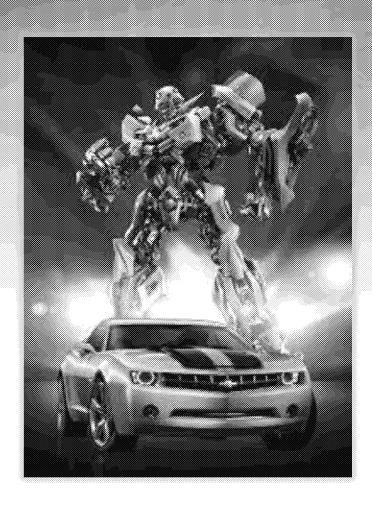
- Risk = one in one million risk of cancer from a lifetime exposure with no adverse effects
- NJ recommended health based MCL based on cancer and noncancer endpoints = 14 ppt
- Production and use of PFOA in U.S. phased out
- Exposure continues due to persistence, biotransformation of precursor compounds and manufacturing abroad





### Polyfluorinated - PFAA Precursors



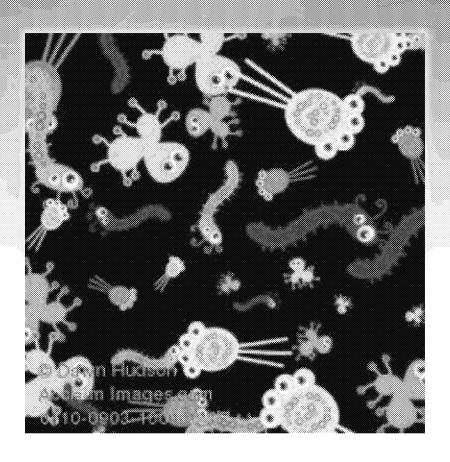


- Thousands of PFAS are used in industrial and consumer products
- Some biotransform to make PFAAs
- Some are fluorotelomers
- Most are ionic either positive, negative or both
- Fate and transport complex process



# How Do Other PFAS Become PFOA?





### Primarily 2 mechanisms:

- Abiotic transformation of PFAA precursors sulfonamido and fluorotelomer precursors oxidize to form PFCAs
- Aerobic biotransformation of fluorotelomer precursors to form PFCAs
- Other biological mechanisms exist

### Discrete PFAAs and Precursors

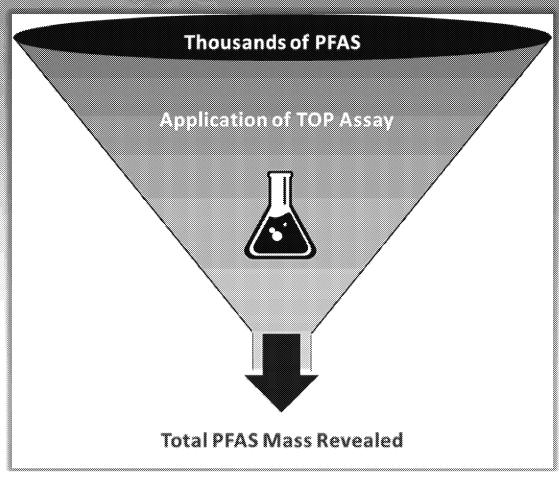


| Compound Name                                    | Abbreviation                         | CAS#       |
|--|--------------------------------------|------------|
| PFAAs - Perfluoroalky                            | lcarboxylic acids (PFCAs)            |            |
| Perfluoro-n-butanoic acid                        | PFBA                                 | 375-22-4   |
| Perfluoro-n-pentanoic acid                       | PFPeA                                | 2706-90-3  |
| Perfluoro-n-hexanoic acid                        | PFHxA                                | 307-24-4   |
| Perfluoro-n-heptanoic acid                       | PFHpA                                | 375-85-9   |
| Perfluoro-n-octanoic acid                        | PFOA                                 | 335-67-1   |
| Perfluoro-n-nonanoic acid                        | PFNA                                 | 375-95-1   |
| Perfluoro-n-decanoic acid                        | PFDA                                 | 335-76-2   |
| Perfluoro-n-undecanoic acid                      | PFUdA                                | 2058-94-8  |
| Perfluoro-n-dodecanoic acid                      | PFDoA                                | 307-55-1   |
| Perfluoro-n-tridecanoic acid                     | PFTrDA                               | 72629-94-8 |
| Perfluoro-n-tetradecanoic acid                   | PFTeDA                               | 376-06-7   |
| Perfluoro-n-hexadecanoic acid                    | PFHxDA                               | 67905-19-5 |
| Perfluoro-n-octadecanoic acid                    | PFODA                                | 16517-11-6 |
| PFAAs - Perfluorinate                            | ed sulfonic acids (PFSAs)            |            |
| Perfluoro-1-butanesulfonic acid                  | PFBS                                 | 375-73-5   |
| Perfluoro-1-hexanesulfonic acid                  | PFHxS                                | 355-46-4   |
| Perfluoro-1-heptanesulfonic acid                 | PFHpS                                | 375-92-8   |
| Perfluoro-1-octanesulfonic acid                  | PFOS                                 | 1763-23-1  |
| Perfluoro-1-decanesulfonic acid                  | PFDS                                 | 335-77-3   |
| Precursors to PFAAs - Perflu                     | uorinated sulfonamides (FOSAs)       |            |
| Perfluoro-1-octanesulfonamide                    | FOSA                                 | 754-91-6   |
| N-ethylperfluoro-1-octanesulfonamide             | EtFOSA                               | 4151-50-2  |
| N-methylperfluoro-1-octanesulfonamide            | MeFOSA                               | 31506-32-8 |
| Precursors to PFAAs - Perfluoring                | ted sulfonamidoacetic acids (FOSAAs) |            |
| N-ethylperfluoro-1-octanesulfonamidoacetic acid  | EtFOSAA                              | 2991-50-6  |
| N-methylperfluoro-1-octanesulfonamidoacetic acid | MeFOSAA                              | 2355-31-9  |
| Precursors to PFAAs - Flu                        | orotelomer sulfonates (FTSs)         |            |
| 1H,1H,2H,2H-perfluorooctane sulfonate (6:2)      | 6:2 FTS                              | 27619-97-2 |
| 1H,1H,2H,2H-perfluorodecane sulfonate (8:2)      | 8:2 FTS                              | 39108-34-4 |

## What is the TOP Assay?



- A new PFAS sample preparation technique
- Conceptually simple chemistry
- ➤ Used in conjunction with 537M (Not 537) combines pre and post oxidation results
- Indicates presence of unidentified PFAS in water, sediment and soil





Houtz, Erika, and David L. Sedlak. 2012. Oxidative conversion as a means of detecting precursors to perfluoroalkyl acids in urban runoff. *Environmental Science and Technology* 46: 9342-9349

### Potential PFAS Transformation





PFOA and PFOS

Quantified

Non-Discrete PFAS Compounds

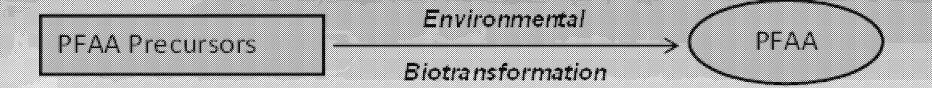
All PFAS Compounds

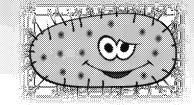
Unknown Potential

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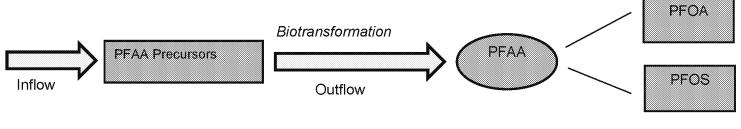
# How Does it Work in the Environment?







#### Give me an example:

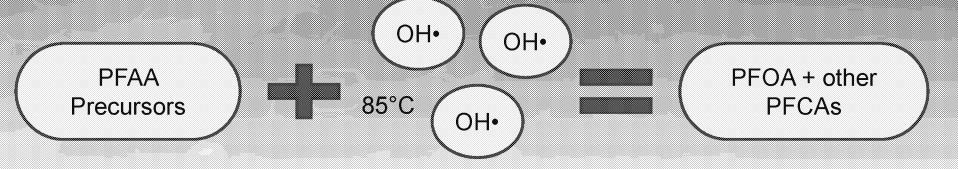


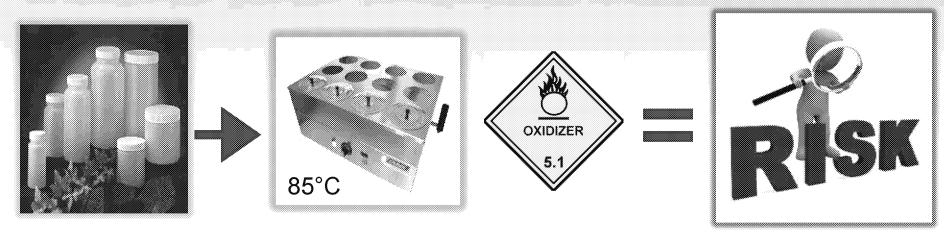
Low levels of discrete compounds are detected

High levels of discrete compounds are detected, which can include PFOA and PFOS

# TOP - How Does it Work in the Laboratory?





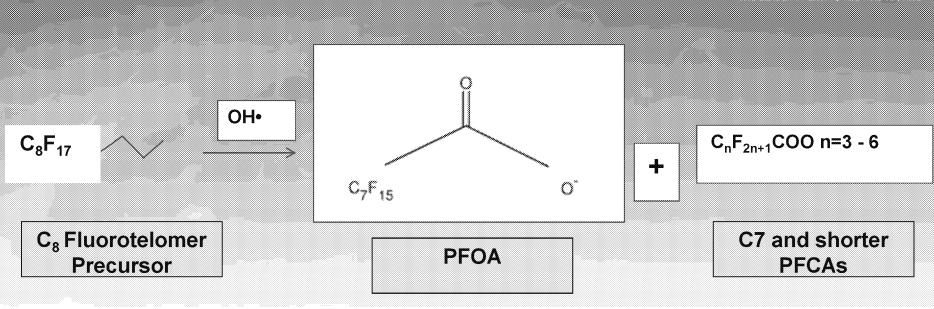


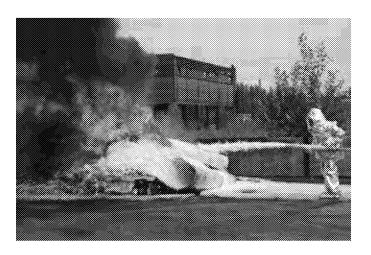


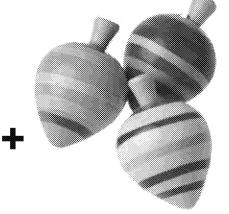
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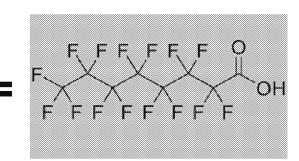
## A Closer Look at the Chemistry











### What Do the Results Mean?



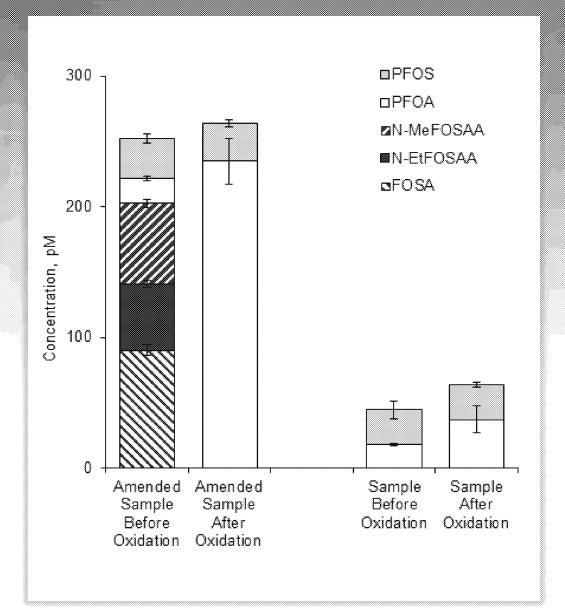
### TOP Assay measures total PFCA

| Precursor | Pre - TOP | Post - TOP | % Oxidation |
|-----------|-----------|------------|-------------|
| ★ FOSA    | 32.68     | ND         | 100%        |
| MeFOSAA   | 19.38     | ND         | 100%        |
| EtFOSAA   | 18.83     | ND         | 100%        |
| 6:2 FTS   | 31.69     | ND         | 100%        |
| 8:2 FTS   | 26.37     | ND         | 100%        |
| PFCA      | Pre – TOP | Post - TOP | Total       |
| PFBA      | 24.94     | 27.16      | 109%        |
| PFPeA     | 23.38     | 28.55      | 122%        |
| PFHxA     | 26.49     | 34.87      | 132%        |
| PFHpA     | 23.10     | 25.14      | 109%        |
|           | 23.72     | 58.71      | 248%        |
|           | Total 122 | Total 174  |             |



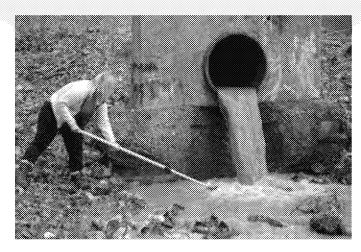
### Urban Runoff - San Jose, CA

### TestAmerico



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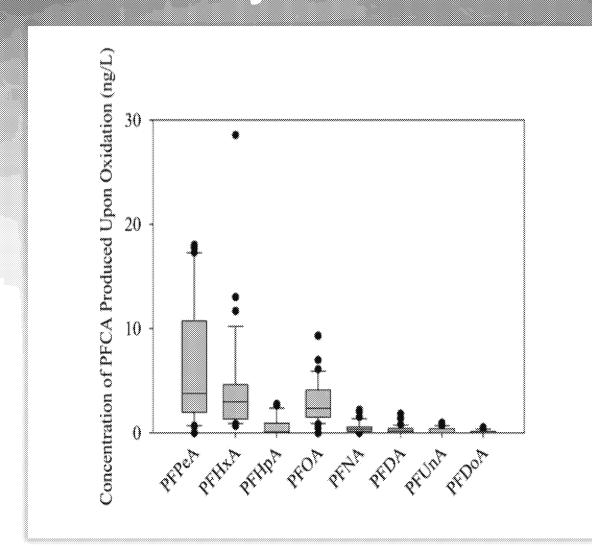


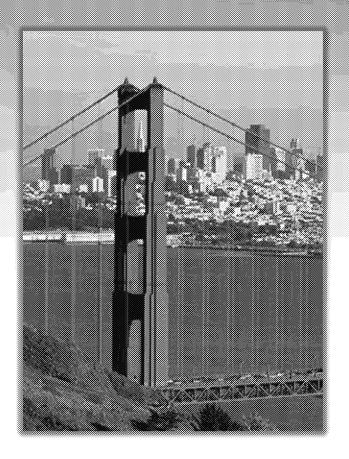


Erika F. Houtz and David L. Sedlak, "Oxidative Conversion as a Means of Detecting Precursors to Perfluoroalkyl Acids in Urban Runoff," Environmental Science and Technology 46, no. 17 (2012): 9342-49.

# Application #1 - Urban Runoff from SF Bay Area



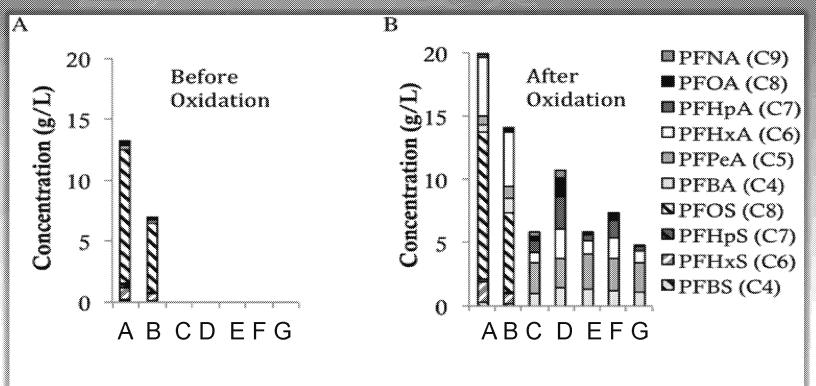




Erika F. Houtz and David L. Sedlak, "Oxidative Conversion as a Means of Detecting Precursors to Perfluoroalkyl Acids in Urban Runoff," Environmental Science and Technology 46, no. 17 (2012): 9342-49.

### App #2 – AFFF Formulations

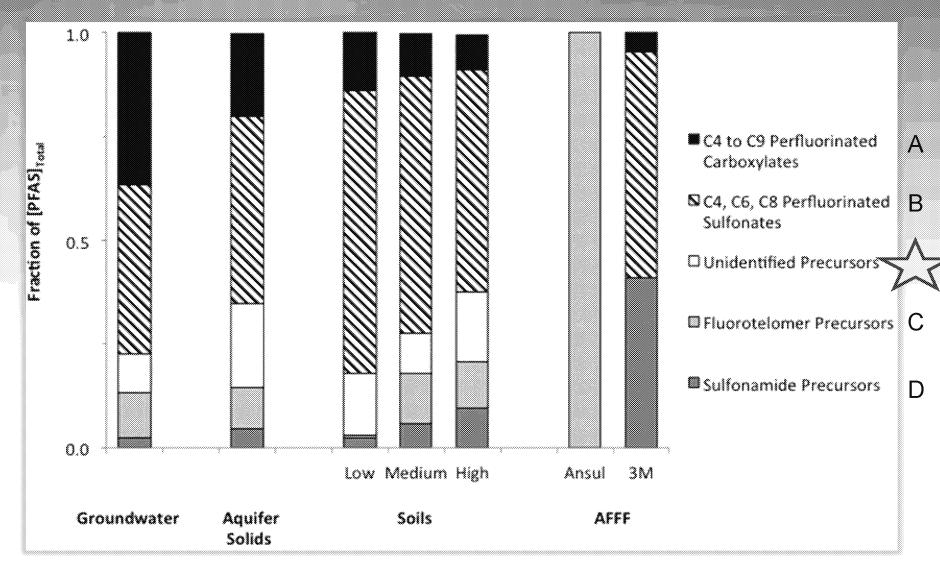




Houtz, E.F., Higgins, C.P., Field, J.A., Sedlak, D.L. Persistence of Perfluoroalkyl Acid Precursors in AFFF-Impacted Groundwater and Soil. Environmental Science & Technology, 2013, 47, 8187-8195.

# Application #3 — AFFF Impacted Groundwater and Soils

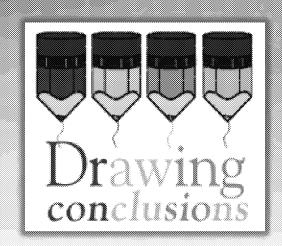




Houtz, E.F., Higgins, C.P., Field, J.A., Sedlak, D.L. Persistence of Perfluoroalkyl Acid Precursors in AFFF-Impacted Groundwater and Soil. Environmental Science & Technology, 2013, 47, 8187-8195.

### What Conclusions Can We Draw?





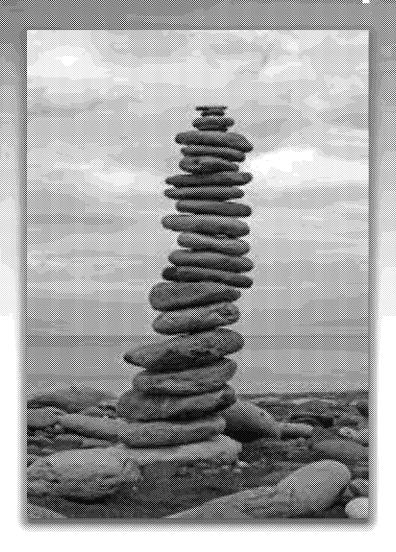
A lesson ...
... in jumping to
CONCLUSIONS



- PFAA precursors are present in environmental samples and many AFFF products
  - Implies treatment strategies must remove precursors and end points
- Presence impacts our treatment strategies and our risk assessments
  - Potentially increases future risk as precursors are biotransformed
- Presence impacts our decisions for AFFF formulations
  - AFFF manufacturers should reduce the content of PFOA etc.

### TestAmerica Capabilities





- TestAmerica Sacramento is EPA approved for Method 537 in drinking water and ISO 25101 in NYS
- Sacramento is QSM 5.1 Table B-15 approved for Method 537M
- Sacramento, Denver and Burlington Labs are NELAP approved for Method 537M.
- 8 LCMSMS instruments capable of PFAS testing
- Sacramento has successfully implemented the TOP Assay

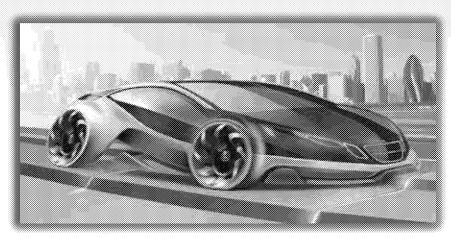


### Future Concerns

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- We need a consensus "best" method.
- Analyte lists are growing for discrete methods, may lead to forensics.
- LC PFASs are being replaced by SC PFASs and little is know about the toxicity
- On-going data variability must be improved
- We need an effective field screening technique.





### References



- Houtz, E.F.; Park, J-S. Sutton, R. Sedlak, M. Poly- and perfluoroalkyl substances in wastewater: Significance of unknown precursors, manufacturing shifts, and likely AFFF impacts. Water Research, 2016, 95, 142-149.
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- McGuire, M.E. Schaefer, C.; Richards, T.; Backe, W.J.; Field, J.A.; Houtz, E.F.; Sedlak, D.L.; Guelfo, J.L.; Wunsch, A.; Higgins, C.P. An In-Depth Site Characterization of Poly- and Perfluoroalkyl Substances at an Abandoned Fire Training Area. *Environmental Science* & *Technology*, 2014, 48, 6644-6652.
- Houtz, E.F., Higgins, C.P., Field, J.A., Sedlak, D.L. Persistence of Perfluoroalkyl Acid Precursors in AFFF-Impacted Groundwater and Soil. *Environmental Science & Technology*, 2013, *47*, 8187-8195.
- Houtz, E.F.; Sedlak. D.L. Oxidative Conversion as a Means of Detecting Precursors to Perfluoroalkyl Acids in Urban Runoff. *Environmental Science & Technology*, 2012, 46, 9342-9349.

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